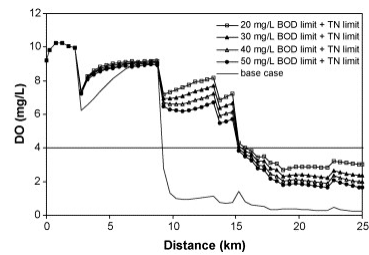
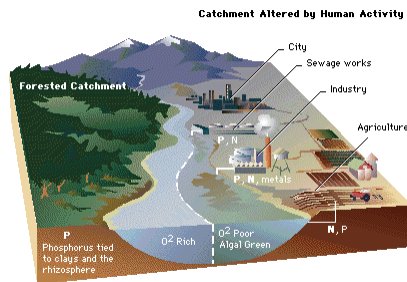


Introduction to Modeling Eutrophication Using QUAL2Kw



Nicholas von Stackelberg, P.E.

December 4, 2012



QUAL2Kw Background

- Developed specifically to simulate dissolved oxygen dynamics in riverine systems
- Widely applied to eutrophication TMDLs and WLAs in the US and internationally
- Developers
 - Dr. Steven Chapra, Tufts University
 - Greg Pelletier, Washington Dept of Ecology
 - Maintained and distributed by Washington Dept of Ecology
- Related Models
 - QUAL2E – Brown and Barnwell 1987
 - QUAL2K – distributed by EPA

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QUAL2Kw Fundamentals

- **Receiving water model** – simulates flow and kinetics of advective systems
- **One dimensional** - channel is uniformly mixed vertically and laterally
- **Steady state flow** - non-uniform, steady flow is simulated
- **Diel meteorology & water quality** – Meteorology and water quality constituents can be varied over 24-hours
- **Flow and water quality inputs** - Point and non-point loads and abstractions can be simulated
- **Water quality kinetics** – temperature, pH, sediment, pathogens, nutrients, organic matter, algal growth and dissolved oxygen

Slide 3



QUAL2Kw Limitations

Does not simulate:

- **Watershed processes**
- **Non-uniform mixing** – 2D or 3D transport
- **Unsteady flow**
Beta version with non-steady, non-uniform flow using kinematic wave flow routing now available
- **Scour and deposition**
- **Adsorption/desorption to sediment**
- **Organisms** – zooplankton, macroinvertebrates, fish
- **Reservoirs**

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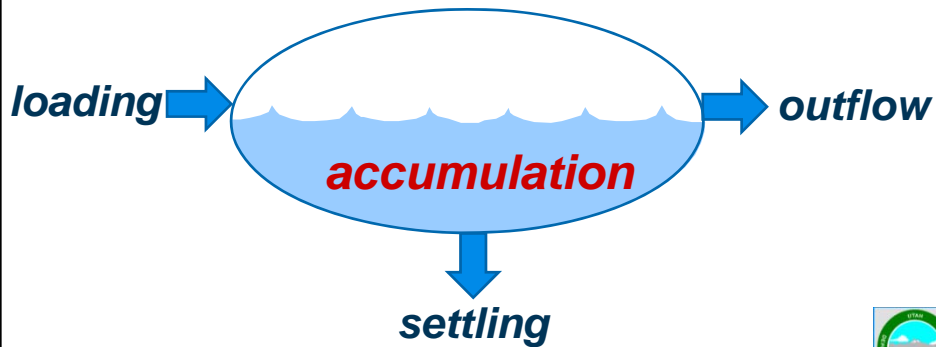


Mechanistic Model

Conservation of Mass

For a finite volume over a unit time period:

$$\text{Accumulation} = \text{Loading} - \text{Outflow} \pm \text{Reaction} - \text{Settling}$$

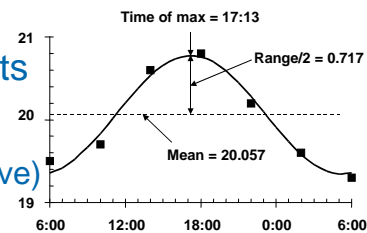


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Model Inputs

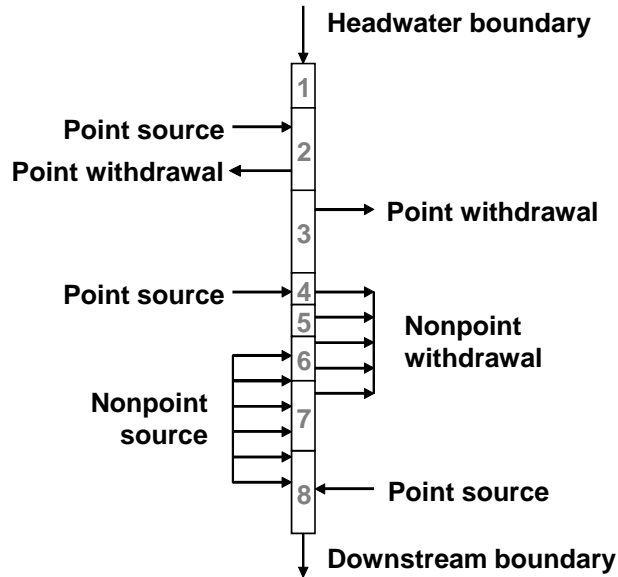
- Dry weather – no precipitation
- Headwaters (hourly)
- Point Sources (sine wave)
 - Tributaries
 - Wastewater treatment plants
 - Industrial discharges
- Point Abstractions (sine wave)
 - Canals/diversions
- Diffuse Sources (sine wave)
 - Groundwater



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Model Segmentation



Slide 7



Hydraulics

Flow (Q) \rightarrow depth (H) and velocity (U)

Given Q : solve for H and U

- Rating Curve
- Mannings Equation for Trapezoidal Channel
- Weir Equation

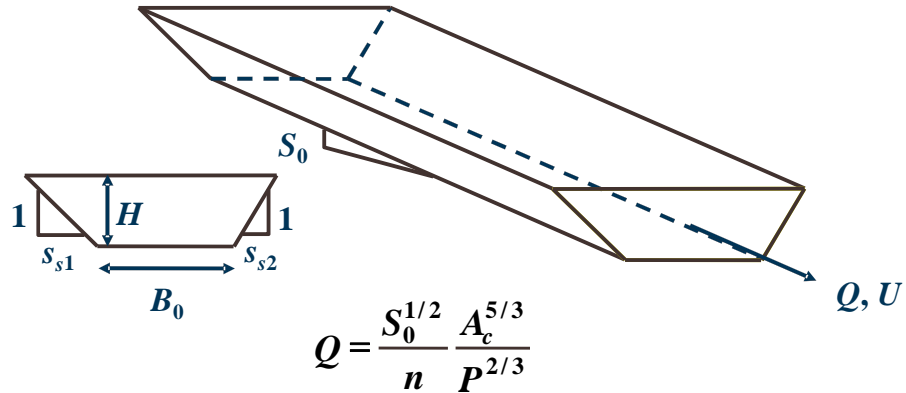
Reach worksheet

Slide 8



Mannings Equation

Trapezoidal Channel



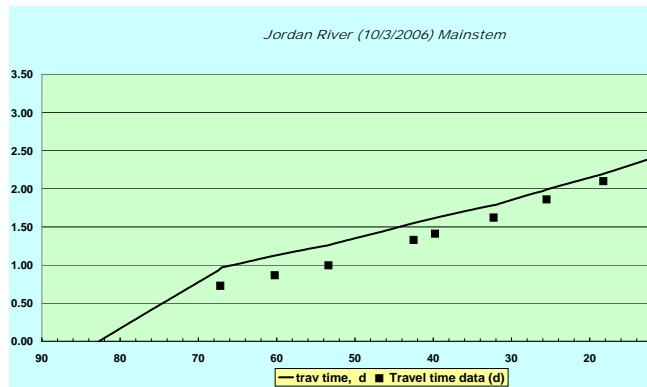
Solves for H and U, given Q, S₀, B₀, s_s



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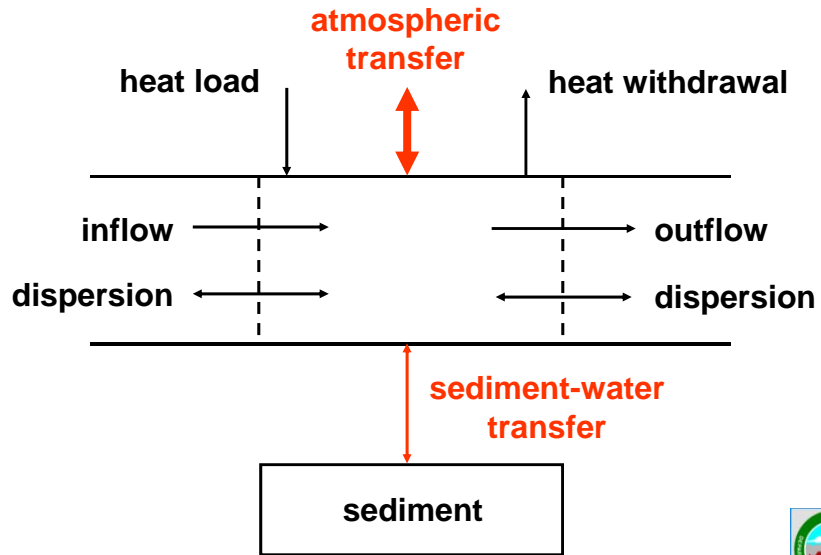
Travel Time

$$\tau_k = \frac{U_k}{Q_k} \quad t_{t,i} = \sum_{k=1}^i \tau_k$$



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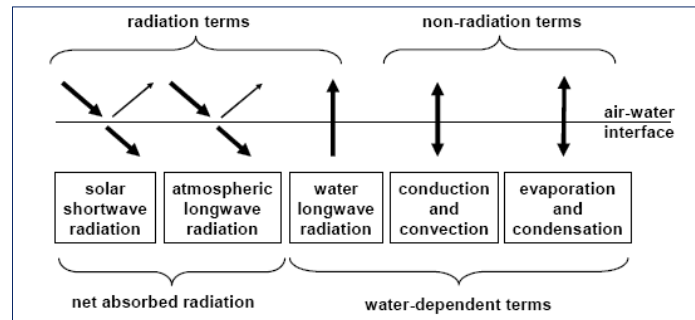
Heat Balance



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Water Surface Heat Balance



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Meteorological Data

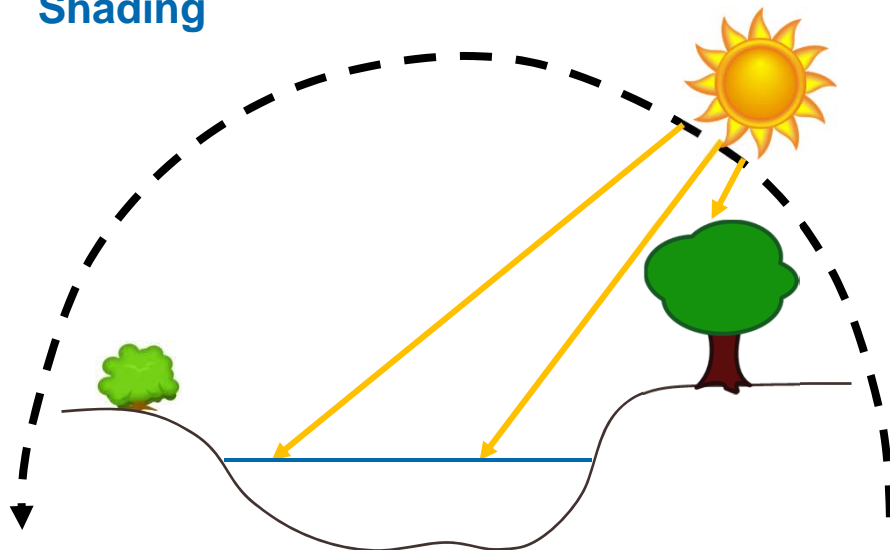
$$J = J_{sn} + \sigma(T_{air} + 273.15)^4 \left(A + 0.031 \sqrt{e_{air}} \right) (1 - R_L) - \varepsilon \sigma (T_s + 273.15)^4 - c_1 f(U_w)(T_s - T_{air}) - f(U_w)(e_s - e_{air})$$

Air Temperature
 Dew Point Temperature
 Solar Radiation (Cloud Cover)
 Wind Speed

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Shading



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Shading Estimation Tools

TTools for ArcGIS

Oregon DEQ

<http://www.deq.state.or.us/WQ/TMDLs/tools.htm>

Shade.xls Excel Spreadsheet

Washington DOE

<http://www.ecy.wa.gov/programs/eap/models.html>



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Water Quality Constituents

Table 5 Model state variables

Variable	Symbol	Units*
Conductivity	s	μmhos
Inorganic suspended solids	m_i	mgD/L
Dissolved oxygen	o	mgO_2/L
Slowly reacting CBOD	c_s	mgO_2/L
Fast reacting CBOD	c_f	mgO_2/L
Organic nitrogen	n_o	$\mu\text{gN/L}$
Ammonia nitrogen	n_a	$\mu\text{gN/L}$
Nitrate nitrogen	n_n	$\mu\text{gN/L}$
Organic phosphorus	p_o	$\mu\text{gP/L}$
Inorganic phosphorus	p_i	$\mu\text{gP/L}$
Phytoplankton	a_p	$\mu\text{gA/L}$
Detritus	m_d	mgD/L
Pathogen	X	$\text{cfu}/100 \text{ mL}$
Alkalinity	Alk	mgCaCO_3/L
Total inorganic carbon	c_T	mole/L
Bottom algae biomass	a_b	mgA/m^2
Bottom algae nitrogen	IN_b	mgN/m^2
Bottom algae phosphorus	IP_b	mgP/m^2

* $\text{mg/L} \equiv \text{g}/\text{m}^3$



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Inorganic Suspended Solids

$$\text{ISS Settling} = \frac{v_i}{H} * \text{ISS}$$

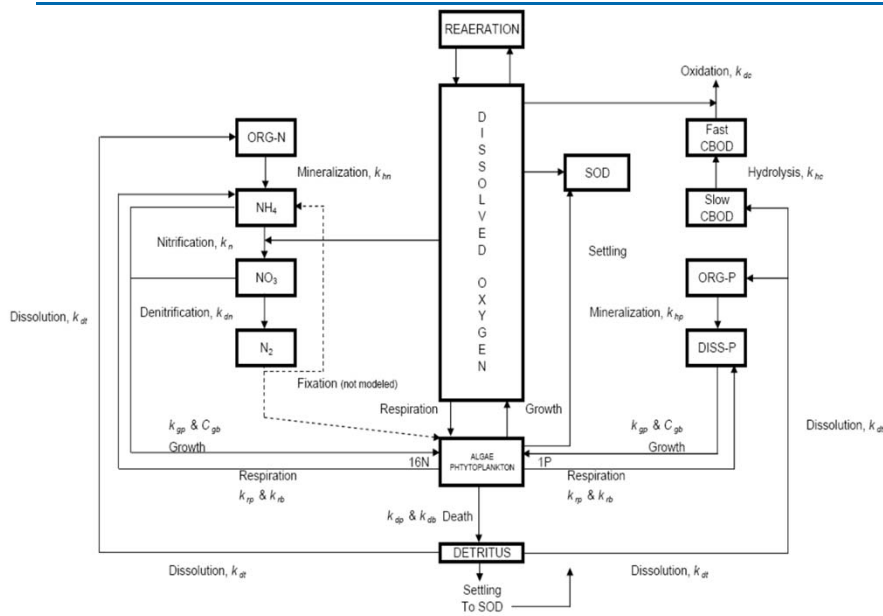
v_i : ISS settling velocity

H : water depth



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Dissolved Oxygen



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Reaeration

Effect of Water Velocity (8 formulas)

17	Reaeration model	USGS(pool-riffle)
18	Temp correction	Internal
19	Reaeration wind effect	O'Connor-Dobbins Churchill
20	O2 for carbon oxidation	Owens-Gibbs Tsivoglou-Neal
21	O2 for NH4 nitrification	Thackston-Dawson USGS(pool-riffle)
22	Oxygen inhib model CBOD oxidation	USGS(channel-control)

Effect of Wind (2 formulas)

19	Reaeration wind effect	None
20	O2 for carbon oxidation	None Banks-Herrera
21	O2 for NH4 nitrification	Wanninkhof

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Soluble Carbonaceous BOD

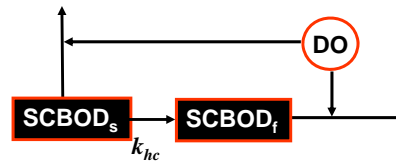
2 Types of SCBOD:

SCBOD_s: slow reacting

SCBOD_f: fast reacting

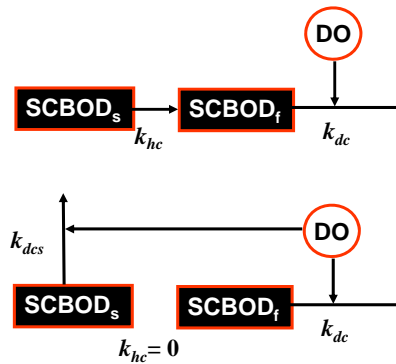
Series:

$$k_{des} = 0$$



Parallel:

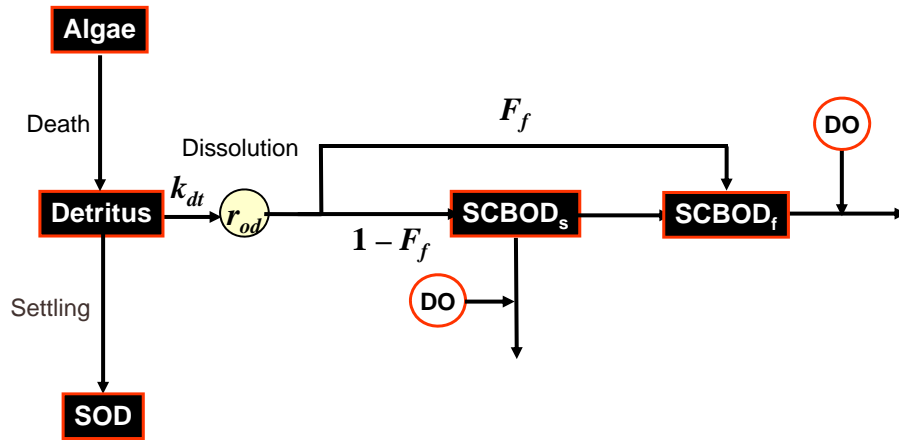
$$k_{hc} = 0$$



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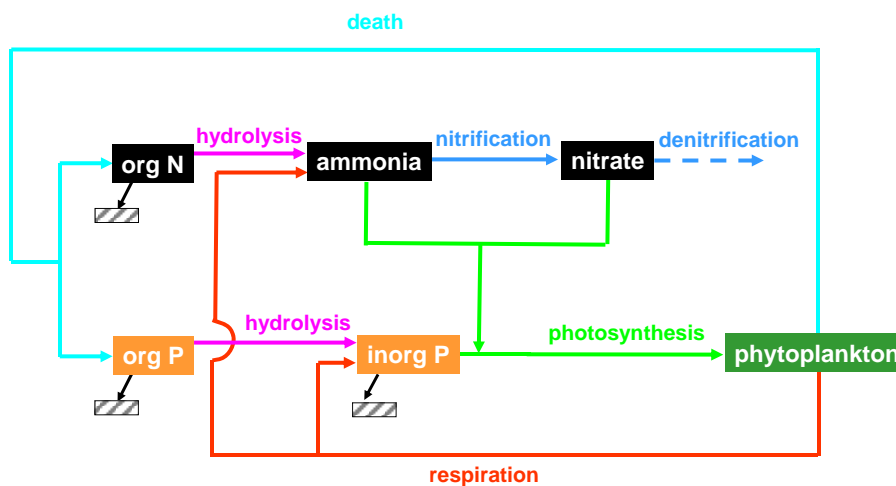
Detritus (Particulate Organic Matter)



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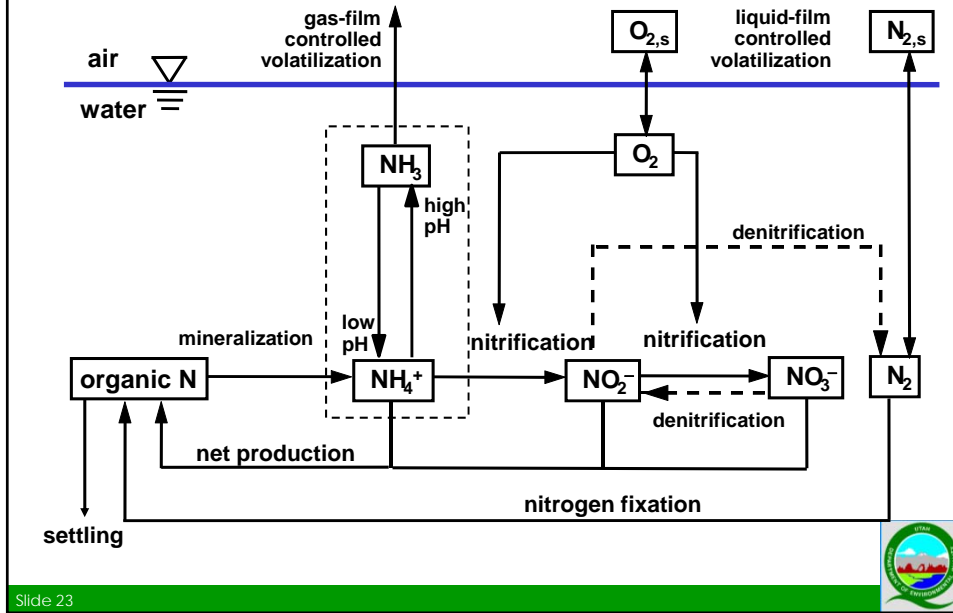
Nutrient Cycle



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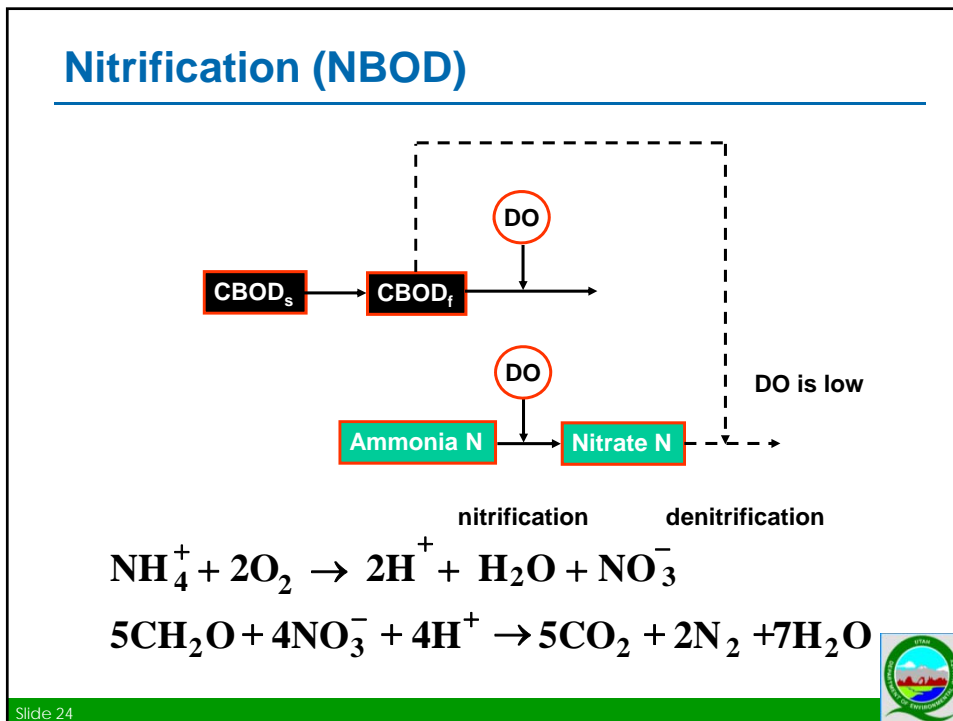


Nitrogen



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Nitrification (NBOD)



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Algal Growth

- Phytoplankton
 - Free floating algae
- Bottom Algae
 - Fixed to stream substrate
 - Periphyton or benthic algae
 - Filamentous algae
 - Macrophytes



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Floating vs. Bottom Plants

	Floating	Bottom
Transport	Yes	No
Types	Diatoms Greens Blue Greens	Periphyton Filamentous Algae Rooted Macrophytes
Units	mgA/m ³	gA/m ²
Light	Ave. Water Column	Bottom Light
Predation	Zooplankton	Insect Larvae, Snails
Substrate	None	Rock vs. Mud



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Phytoplankton Kinetics

$$\frac{da}{dt} = \text{transport} + k_g(T, N, I) a - k_r(T) a - k_d(T) a - (v_s/H) a$$

$k_g(T, N, I) = k_{g,T} \phi_N \phi_L$

inorganic nutrients \uparrow respiration \downarrow death \downarrow detritus organic nutrients \uparrow
 growth \swarrow settling \downarrow sediments

temperature \nearrow nutrients \nearrow light \nearrow

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Phytoplankton Stoichiometry

The "Redfield Ratio"

D : C : N : P : A

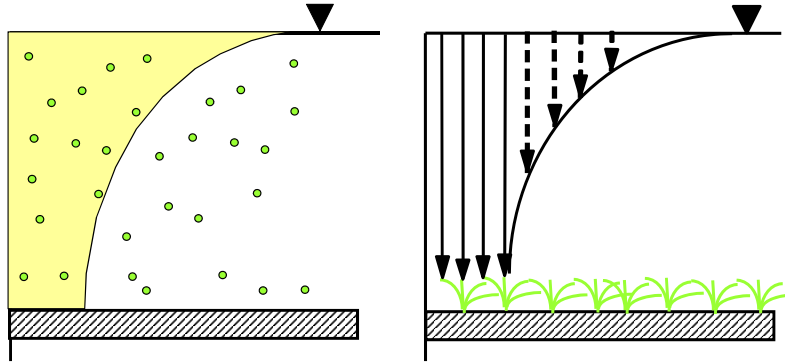
100% : 40% : 7.2% : 1% : 0.5-2.0%

	A	B	C	D
1	QUAL2K			
2	Stream Water Quality Model			
3	Streeter River (8/15/2002)			
4	Water Column Rates			
5				
6				
7	<i>Parameter</i>	<i>Value</i>	<i>Units</i>	<i>Symbol</i>
8	Stoichiometry:			
9	Carbon	40	mgC	
10	Nitrogen	7.2	mgN	
11	Phosphorus	1	mgP	
12	Dry weight	100	mgD	
13	Chlorophyll	1	mgA	

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Light Estimation



(a) floating plants

(b) bottom algae

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Periphyton Growth Rates

Zero-order:

$$C_{gb}(T) = mgA/m^2/d$$

$$\frac{da_b}{dt} = C_{gb}(T)\phi_{Nb}\phi_{Lb}$$



First-order:

$$k_{gb}(T)a_b = mgA/m^2/d$$

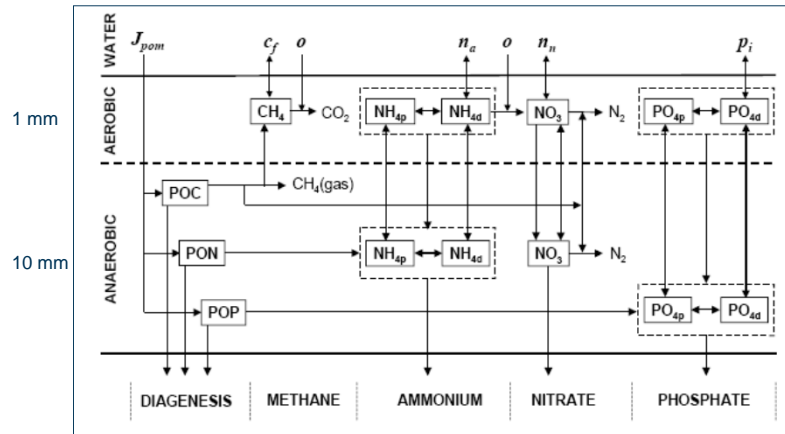
$$\frac{da_b}{dt} = k_{gb}(T)\phi_{Nb}\phi_{Lb}a_b$$



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Sediment Oxygen Demand



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References

- QUAL2K2w User Manual (Version 5.1)*, Greg Pelletier and Steve Chapra, 2008.
- QUAL2K2w Theory and Documentation (Version 5.1)*, Greg Pelletier and Steve Chapra, 2008.
- Hydrodynamics and Water Quality: Modeling Rivers, Lakes and Estuaries*, Zheng-Gang Ji, John Wiley and Sons, 2008.
- Surface Water Quality Modeling*, Steven Chapra, WCB McGraw Hill, 1997.
- Principles of Surface Water Quality Modeling and Control*, R.V. Thomann and J.A. Mueller, Harper and Row, 1987.
- Sediment Flux Modeling*, Dominic DiToro, John Wiley and Sons, 2001.
- The Enhanced Stream Water Quality Models QUAL2E and QUAL2E-UNCAS: Documentation and Users Manual*, L.C. Brown and T.O. Barnwell, EPA/600/3-87-007, 1987.

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